You might think that you can tell if a body of water is polluted by the way that the water looks or smells, but sometimes you can’t. There are many different forms of water pollution. **Water pollution** is the introduction of chemical, physical, or biological agents into water that degrade water quality and adversely affect the organisms that depend on the water. Almost all of the ways that we use water contribute to water pollution. However, the two underlying causes of water pollution are industrialization and rapid human population growth.

In the last 30 years, developed countries have made great strides in cleaning up many polluted water supplies. Despite this progress, some water is still dangerously polluted in the United States and in other countries. In developing parts of the world, water pollution is a big problem. Industry is usually not the major cause of water pollution in developing countries. Often, the only water available for drinking in these countries is polluted with sewage and agricultural runoff, which can spread waterborne diseases. To prevent water pollution, people must understand where pollutants come from. As you will learn, water pollution comes from two types of sources: point and nonpoint sources.
**Point-Source Pollution**

When you think of water pollution, you probably think of a single source, such as a factory, a wastewater treatment plant, or a leaking oil tanker. These are all examples of **point-source pollution**, which is pollution discharged from a single source. **Table 3** lists some additional examples of point-source pollution. Point-source pollution can often be identified and traced to a source. But even when the source of pollution is known, enforcing cleanup is sometimes difficult.

**Nonpoint-Source Pollution**

Nonpoint-source pollution comes from many different sources that are often difficult to identify. For example, a river can be polluted by runoff from any of the land in its watershed. If a farm, a road, or any other land surface in a watershed is polluted, runoff from a rainstorm can carry the pollution into a nearby river, stream, or lake. **Figure 16** shows common sources of nonpoint pollutants. **Table 4** lists some additional causes of nonpoint pollution.

Because nonpoint pollutants can enter bodies of water in many different ways, they are extremely difficult to regulate and control. The accumulation of small amounts of water pollution from many sources is a major pollution problem—96 percent of the polluted bodies of water in the United States were contaminated by nonpoint sources. Controlling nonpoint-source pollution depends to a great extent on public awareness of the effects of activities such as spraying lawn chemicals and using storm drains to dispose of used motor oil.

---

**Table 4**

<table>
<thead>
<tr>
<th>Nonpoint Sources of Pollution</th>
</tr>
</thead>
<tbody>
<tr>
<td>• chemicals added to road surfaces (salt and other de-icing agents)</td>
</tr>
<tr>
<td>• water runoff from city and suburban streets that may contain oil, gasoline, animal feces, and litter</td>
</tr>
<tr>
<td>• pesticides, herbicides, and fertilizer from residential lawns, golf courses, and farmland</td>
</tr>
<tr>
<td>• feces and agricultural chemicals from livestock feedlots</td>
</tr>
<tr>
<td>• precipitation containing air pollutants</td>
</tr>
<tr>
<td>• soil runoff from farms and construction sites</td>
</tr>
<tr>
<td>• oil and gasoline from personal watercraft</td>
</tr>
</tbody>
</table>

**Figure 16**  
**Sources of Nonpoint Pollution** Examples of nonpoint-source pollution include 1 livestock polluting water holes that can flow into streams and reservoirs, 2 oil on a street, which can wash into storm sewers and then drain into waterways, and 3 thousands of watercraft, which can leak gasoline and oil.

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Principal Water Pollutants

There are many different kinds of water pollutants. Table 5 lists some common types of pollutants and some of the possible sources of each pollutant.

Wastewater

Do you know where water goes after it flows down the drain in a sink? The water usually flows through a series of sewage pipes that carry it—and all the other wastewater in your community—to a wastewater treatment plant. **Wastewater** is water that contains waste from homes or industry. At a wastewater treatment plant, water is filtered and treated to make the water clean enough to return to a river or lake.

Treating Wastewater  A typical residential wastewater treatment process is illustrated in Figure 17. Most of the wastewater from homes contains biodegradable material that can be broken down by living organisms. For example, wastewater from toilets and kitchen sinks contains animal and plant wastes, paper, and soap, all of which

<table>
<thead>
<tr>
<th>Pollutant Types and Sources</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Type of pollutant</strong></td>
</tr>
<tr>
<td>Pathogens</td>
</tr>
<tr>
<td>Organic matter</td>
</tr>
<tr>
<td>Organic chemicals</td>
</tr>
<tr>
<td>Inorganic chemicals</td>
</tr>
<tr>
<td>Heavy metals</td>
</tr>
<tr>
<td>Physical agents</td>
</tr>
</tbody>
</table>
are biodegradable. But wastewater treatment plants may not remove all of the harmful substances in water. Some household and industrial wastewater and some storm-water runoff contains toxic substances that cannot be removed by the standard treatment.

**Sewage Sludge** If you look again at Figure 17, you will see that one of the products of wastewater treatment is sewage sludge, the solid material that remains after treatment. When sludge contains dangerous concentrations of toxic chemicals, it must be disposed of as hazardous waste. The sludge is often incinerated, and then the ash is buried in a secure landfill. Sludge can be an expensive burden to towns and cities because the volume of sludge that has to be disposed of every year is enormous.

The problem of sludge disposal has prompted many communities to look for new uses for this waste. If the toxicity of sludge can be reduced to safe levels, sludge can be used as a fertilizer. In another process, sludge is combined with clay to make bricks that can be used in buildings. In the future, industries will probably find other creative ways to use sludge.

---

**Connection to History**

**Cryptosporidium Outbreak** In 1993, a pathogen called *Cryptosporidium parvum* contaminated the municipal water supply of Milwaukee, Wisconsin. The waterborne parasite caused more than 100 deaths, and 400,000 people experienced a flulike illness. *Cryptosporidium* is found in animal feces, but the parasite usually occurs in low levels in water supplies. The outbreak in Milwaukee was probably caused by an unusual combination of heavy rainfall and agricultural runoff that overburdened the city’s water treatment plants.

---

**Figure 17**  
**Wastewater Treatment Process**

**PRIMARY TREATMENT**

1. **Filtration**  Wastewater is passed through a large screen to remove solid objects.

2. **Settling Tank**  Wastewater is sent into a large tank, where smaller particles sink to the bottom and form sewer sludge. The sludge is removed from the water.

---

**SECONDARY TREATMENT**

3. **Second Filtration**  Wastewater is sent to a large tank, where any remaining sludge is removed from the water.

4. **Aeration Tank**  Wastewater is mixed with oxygen and bacteria. The bacteria use the oxygen and feed on the wastes.

5. **Chlorination**  Chlorine is added to disinfect the water before it is released into a stream, lake, or ocean.
QuickLAB

Measuring Dissolved Oxygen

Procedure
1. Start with three water samples, each in a plastic jar that is \( \frac{3}{4} \) full. Two water samples should be tap water from a faucet without an aerator. One sample should be water that has been boiled and allowed to cool.
2. Using a dissolved-oxygen test kit, test the boiled water and one other water sample.
3. Tighten the lid on the third sample, and then vigorously shake the sample for one minute. Unscrew the lid, and then recap the jar.
4. Repeat step 3 twice. Then, uncap the jar quickly, and test the sample.

Analysis
1. Which sample had the highest dissolved oxygen level? Which sample had the lowest level?
2. What effects do rapids and waterfalls have on the levels of dissolved oxygen in a stream? What effect does thermal pollution have?

Artificial Eutrophication
Most nutrients in water come from organic matter, such as leaves and animal waste, that is broken down into mineral nutrients by decomposers such as bacteria and fungi. Nutrients are an essential part of any aquatic ecosystem, but an overabundance of nutrients can disrupt an ecosystem. When lakes and slow-moving streams contain an abundance of nutrients, they are eutrophic (yoo TROH fi).

Eutrophication is a natural process. When organic matter builds up in a body of water, it will begin to decay and decompose. The process of decomposition uses up oxygen. As oxygen levels decrease, the types of organisms that live in the water change over time. For example, as a body of water becomes eutrophic, plants take root in the nutrient-rich sediment at the bottom. As more plants grow, the shallow waters begin to fill in. Eventually, the body of water becomes a swamp or marsh.

The natural process of eutrophication is accelerated when inorganic plant nutrients, such as phosphorus and nitrogen, enter the water from sewage and fertilizer runoff. Eutrophication caused by humans is called artificial eutrophication. Fertilizer from farms, lawns, and gardens is the largest source of nutrients that cause artificial eutrophication. Phosphates in some laundry and dishwashing detergents are another major cause of eutrophication. Phosphorus is a plant nutrient that can cause the excessive growth of algae. In bodies of water polluted by phosphorus, algae can form large floating mats, called algal blooms, as shown in Figure 18. As the algae die and decompose, most of the dissolved oxygen is used and fish and other organisms suffocate in the oxygen-depleted water.

![Figure 18](image-url) In an effort to limit artificial eutrophication, some states have either banned phosphate detergents or limited the amount of phosphates in detergents.
Thermal Pollution

If you look at Figure 19, you might assume that a toxic chemical caused the massive fish kill in the photo. But the fish were not killed by a chemical spill—they died because of thermal pollution. When the temperature of a body of water, such as a lake or stream, increases, thermal pollution can result. Thermal pollution can occur when power plants and other industries use water in their cooling systems and then discharge the warm water into a lake or river.

Thermal pollution can cause large fish kills if the discharged water is too warm for the fish to survive. But most thermal pollution is more subtle. If the temperature of a body of water rises even a few degrees, the amount of oxygen the water can hold decreases significantly. As oxygen levels drop, aquatic organisms may suffocate and die. If the flow of warm water into a lake or stream is constant, it may cause the total disruption of an aquatic ecosystem.

Groundwater Pollution

Pollutants usually enter groundwater when polluted surface water percolates down from the Earth’s surface. Any pollution of the surface water in an area can affect the groundwater. Pesticides, herbicides, chemical fertilizers, and petroleum products are common groundwater pollutants. Leaking underground storage tanks are another major source of groundwater pollution. It is estimated that there are millions of underground storage tanks in the United States. Most of the tanks—located beneath gas stations, farms, and homes—hold petroleum products, such as gasoline and heating fuel. As underground storage tanks age, they may develop leaks, which allow pollutants to seep into the groundwater.

Figure 19 ▶ Fish kills, such as this one in Brazil, can result from thermal pollution.

Connection to Chemistry

Dissolved Oxygen One of the most important measures of the health of a body of water is the amount of dissolved oxygen in the water. Gaseous oxygen enters water by diffusion from the surrounding air, as a byproduct of photosynthesis, and as a result of the rapid movement (aeration) of water. The amount of oxygen that water can hold is determined by the water’s temperature, pressure, and salinity. Slow-moving waters tend to have low levels of dissolved oxygen, while rapidly flowing streams have higher levels. Artificial eutrophication and thermal pollution also reduce levels of dissolved oxygen. When dissolved oxygen levels remain below 1 to 2 mg/L for several hours, fish and other organisms suffocate, and massive fish kills can result.
The location of aging underground storage tanks is not always known, so the tanks often cannot be repaired or replaced until after they have leaked enough pollutants to be located. Modern underground storage tanks are contained in concrete and have many features to prevent leaks. Other sources of groundwater pollution include septic tanks, unlined landfills, and industrial wastewater lagoons, as shown in Figure 20.

**Cleaning Up Groundwater Pollution**  Groundwater pollution is one of the most challenging environmental problems that the world faces. Even if groundwater pollution could be stopped tomorrow, some groundwater would remain polluted for generations to come. As you have learned, groundwater recharges very slowly. The process for some aquifers to recycle water and purge contaminants can take hundreds or thousands of years. Groundwater is also difficult to decontaminate because the water is dispersed throughout large areas of rock and sand. Pollution can cling to the materials that make up an aquifer, so even if all of the water in an aquifer were pumped out and replaced with clean water, the groundwater could still become polluted.

Figure 20  This diagram shows some of the major sources of groundwater pollution. Runoff and percolation transport contaminants to the groundwater.
Ocean Pollution

Although oceans are the largest bodies of water on Earth, they are still vulnerable to pollution. Pollutants are often dumped directly into the oceans. For example, ships can legally dump wastewater and garbage overboard in some parts of the ocean. But at least 85 percent of ocean pollution—including pollutants such as oil, toxic wastes, and medical wastes—comes from activities on land. If polluted runoff enters rivers, for example, the rivers may carry the polluted water to the ocean. Most activities that pollute oceans occur near the coasts, where much of the world’s human population lives. As you might imagine, sensitive coastal ecosystems, such as coral reefs, estuaries, and coastal marshes, are the most affected by pollution.

Oil Spills  Ocean water is also polluted by accidental oil spills. Disasters such as the 1989 Exxon Valdez oil spill in Prince William Sound, Alaska, make front-page news around the world. In 2001, a fuel-oil spill off the coast of the Galápagos Islands captured public attention. Each year, approximately 37 million gallons of oil from tanker accidents are spilled into the oceans. Figure 21 shows some of the major oil spills that occurred off the coast of North America in the last 30 years.

Such oil spills have dramatic effects, but they are responsible for only about 5 percent of oil pollution in the oceans. Most of the oil that pollutes the oceans comes from cities and towns. Every year, as many as 200 million to 300 million gallons of oil enter the ocean from nonpoint sources on land. That’s almost 10 times the amount of oil spilled by tankers. In fact, in one year, the road runoff from a coastal city of 5 million people could contain as much oil as a tanker spill does. Limiting these nonpoint sources of oil pollution would go a long way toward keeping the oceans clean.
Water Pollution and Ecosystems

Water pollution can cause immediate damage to an ecosystem. For example, toxic chemicals spilled directly into a river can kill nearly all living things for miles downstream. But the effects of water pollution can be even more far reaching. Many pollutants accumulate in the environment because they do not decompose quickly. As the pollutant levels increase, they can threaten an entire ecosystem.

Consider a river ecosystem. Soil tainted with pesticides washes into the river and settles to the river bottom. Some of the pesticides enter the bodies of tiny, bottom-dwelling organisms, such as insect larvae and crustaceans. A hundred of these organisms are eaten by one small fish. A hundred of these small fish are eaten by one big fish. A predatory bird, such as an eagle, eats 10 big fish. Each organism stores the pesticide in its tissues, so at each step along the food chain, the amount of the pesticide passed on to the next organism increases. This accumulation of pollutants at successive levels of the food chain is called biomagnification. Biomagnification, which is illustrated in Figure 22, has alarming consequences for organisms at the top of the food chain. Biomagnification is one reason why many U.S. states limit the amount of fish that people can eat from certain bodies of water.

Cleaning Up Water Pollution

In 1969, the Cuyahoga River in Cleveland, Ohio, was so polluted that the river caught on fire and burned for several days, as shown in Figure 23. This shocking event was a major factor in the passage of the Clean Water Act of 1972. The stated purpose of the act was to “restore and maintain the chemical, physical, and biological integrity of the nation’s waters.” The goal of the act was to make all surface water clean enough for fishing and swimming by 1983. This goal was not achieved; however, much progress has been made since the act was passed. The percentage of lakes and rivers that are fit for swimming and fishing has increased by about 30 percent, and many states have passed stricter water-quality standards of their own. Many toxic metals are now removed from wastewater before the water is discharged.

The Clean Water Act opened the door for other water-quality legislation, some of which is described in Table 6. For example, the Marine Protection, Research, and Sanctuaries Act of 1972 strengthened the laws against ocean dumping.

The Oil Pollution Act of 1990 requires all oil tankers traveling in U.S. waters to
have double hulls by 2015 as an added protection against oil spills. Legislation has improved water quality in the United States, but the cooperation of individuals, businesses, and the government will be essential to maintaining a clean water supply in the future.

### Table 6

<table>
<thead>
<tr>
<th>Federal Laws Designed to Improve Water Quality in the United States</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>1972 Clean Water Act (CWA)</strong> The CWA set a national goal of making all natural surface water fit for fishing and swimming by 1983 and banned pollutant discharge into surface water after 1985. The act also required that metals be removed from wastewater.</td>
</tr>
<tr>
<td><strong>1972 Marine Protection, Research, and Sanctuaries Act, amended 1988</strong> This act empowered the EPA to control the dumping of sewage wastes and toxic chemicals in U.S. waters.</td>
</tr>
<tr>
<td><strong>1975 Safe Drinking Water Act (SDWA), amended 1996</strong> This act introduced programs to protect groundwater and surface water from pollution. The act emphasized sound science and risk-based standards for water quality. The act also empowered communities in the protection of source water, strengthened public right-to-know laws, and provided water system infrastructure assistance.</td>
</tr>
<tr>
<td><strong>1980 Comprehensive Environmental Response Compensation and Liability Act (CERCLA)</strong> This act is also known as the Superfund Act. The act makes owners, operators, and customers of hazardous waste sites responsible for the cleanup of the sites. The act has reduced the pollution of groundwater by toxic substances leached from hazardous waste dumps.</td>
</tr>
<tr>
<td><strong>1987 Water Quality Act</strong> This act was written to support state and local efforts to clean polluted runoff. It also established loan funds to pay for new wastewater treatment plants and created programs to protect major estuaries.</td>
</tr>
<tr>
<td><strong>1990 Oil Pollution Act</strong> This act attempts to protect U.S. waterways from oil pollution by requiring that oil tankers in U.S. waters be double-hulled by 2015.</td>
</tr>
</tbody>
</table>

### SECTION 3 Review

1. **Explain** why point-source pollution is easier to control than nonpoint-source pollution.

2. **List** the major types of water pollutants. Suggest ways to reduce the levels of each type of pollutant in a water supply.

3. **Describe** the unique problems of cleaning up groundwater pollution.

4. **Describe** the source of most ocean pollution. Is it point-source pollution or nonpoint-source pollution?

### CRITICAL THINKING

5. **Interpreting Graphics** Read the description of biomagnification. Draw a diagram that shows the biomagnification of a pollutant in an ecosystem.

6. **Applying Ideas** What can individuals do to decrease ocean pollution? Write and illustrate a guide that gives at least three examples.
CHAPTER 11

Highlights

1 Water Resources

**Key Terms**
- surface water, 270
- river system, 271
- watershed, 271
- groundwater, 272
- aquifer, 272
- porosity, 273
- permeability, 273
- recharge zone, 274

**Main Ideas**
- Only a small fraction of Earth’s water supply is fresh water. The two main sources of fresh water are surface water and groundwater.
- River systems drain the land that makes up a watershed. The amount of water in a river system can vary in different seasons and from year to year.
- Groundwater accumulates in underground formations called aquifers. Surface water enters an aquifer through the aquifer’s recharge zone.
- If the water in an aquifer is pumped out faster than it is replenished, the water table drops, which can affect humans and animals that depend on the groundwater.

2 Water Use and Management

**Key Terms**
- potable, 277
- pathogen, 277
- irrigation, 278
- dam, 280
- reservoir, 280
- desalination, 283

**Main Ideas**
- There are three main types of water use: residential, industrial, and agricultural. Worldwide, most water use is agricultural.
- Dams and water diversion projects are built to manage surface-water resources. Damming and diverting rivers can have environmental and social consequences.
- Water conservation is necessary to maintain an adequate supply of fresh water. Desalination and transporting water are options to supplement local water supplies.

3 Water Pollution

**Key Terms**
- water pollution, 284
- point-source pollution, 285
- nonpoint-source pollution, 285
- wastewater, 286
- artificial eutrophication, 288
- thermal pollution, 289
- biomagnification, 292

**Main Ideas**
- Water can become polluted by chemical, physical, or biological agents. Most water pollution in the United States is caused by nonpoint-source pollutants.
- Groundwater pollution is difficult to clean up because aquifers recharge slowly and because pollutants cling to the materials that make up an aquifer.
- Ocean pollution is mainly caused by coastal, nonpoint-source pollutants.
- Government legislation, such as the Clean Water Act of 1972, has succeeded in reducing surface-water pollution. Future challenges include reducing nonpoint-source pollution and groundwater pollution.
Using Key Terms

Use each of the following terms in a separate sentence.

1. aquifer
2. recharge zone
3. irrigation
4. wastewater
5. biomagnification

For each pair of terms, explain how the meanings of the terms differ.

6. surface water and groundwater
7. porosity and permeability
8. watershed and river system
9. point-source pollution and nonpoint-source pollution

10. Which of the following processes is not a part of the water cycle?
    a. evaporation
    b. condensation
    c. biomagnification
    d. precipitation

11. Most of the fresh water on Earth is
    a. located underground in aquifers.
    b. frozen in the polar icecaps.
    c. located in rivers, lakes, streams, and wetlands.
    d. found in Earth’s atmosphere.

12. Which of the following processes is not used in a conventional method of water treatment?
    a. filtration
    b. coagulation
    c. aeration
    d. percolation

13. Which of the following is not an example of point-source pollution?
    a. oil that is escaping from a damaged tanker
    b. heavy metals that are leaching out of an underground mine
    c. water runoff from residential lawns
    d. untreated sewage that is accidentally released from a wastewater treatment plant

14. Which of the following pollutants causes artificial eutrophication?
    a. heavy metals from unlined landfills
    b. inorganic plant nutrients from wastewater and fertilizer runoff
    c. toxic chemicals from factories
    d. radioactive waste from nuclear power plants

15. Pumping large amounts of water from an aquifer may cause the
    a. water table to rise.
    b. recharge zone to shrink.
    c. wells in an area to run dry.
    d. percolation of groundwater to stop.

16. Oil pollution in the ocean is mostly caused by
    a. major oil spills, such as the 1989 Exxon Valdez oil spill.
    b. the cumulative effect of small oil spills and leaks on land.
    c. decomposed plastic materials.
    d. intentional dumping of excess oil.

17. Thermal pollution has a harmful effect on aquatic environments because
    a. water has been circulated around power-plant generators.
    b. it increases the number of disease-causing organisms in aquatic environments.
    c. it reduces the amount of dissolved oxygen in aquatic environments.
    d. it decreases the nutrient levels in aquatic environments.
Short Answer

18. What effect can buildings and parking lots have on an aquifer’s recharge zone?
19. Why is the use of overhead sprinklers for irrigation inefficient? What is a more efficient method of irrigation?
20. List three advantages and three disadvantages of dams.
21. What is the process of eutrophication, and how do human activities accelerate it?
22. Describe the steps that are involved in the primary and secondary treatment of wastewater.

Interpreting Graphics

The graph below shows the annual flow, or discharge, of the Yakima River in Washington. Use the graph to answer questions 23–25.

23. In which months is the river’s discharge highest? What might explain these discharge rates?
24. What might cause the peaks in river discharge between November and March?
25. How might the data be different if the hydrograph readings were taken below a large dam on the Yakima River?

Concept Mapping

26. Use the following terms to create a concept map: Earth’s surface, rivers, underground, fresh water, water table, 3 percent, and icecaps.

Critical Thinking

27. Making Comparisons Read the description of artificial eutrophication in this chapter. Do you think artificial eutrophication is more disturbing to the stability of a water ecosystem than natural eutrophication is?

28. Analyzing Relationships Water resources are often shared by several countries. A river, for example, might flow through five countries before it reaches an ocean. When water resources are shared, how should countries determine water rights and environmental responsibility?

29. Making Inferences Explain why it takes 36 gallons of water to produce a single serving of rice, but it takes more than 2,000 gallons of water to produce a single serving of steak. What do you think the water is used for in each case?

30. Making Inferences Why is there so little fresh water in the world? Do you think that there would have been more fresh water at a different time in Earth’s history?

Cross-Disciplinary Connection

31. Social Studies Find out how freshwater resources affected the development of one culture in history. Use at least five key terms from this chapter to write a two-paragraph description of how the availability of fresh water affected the culture you chose.

Portfolio Project

32. Investigation Find out about the source of the tap water in your home. Where does the tap water come from, and where does your wastewater go? Does the water complete a cycle? Make a poster to illustrate your findings. You may want to work with several classmates and visit the sites you discover.
Read the passage below, and then answer the questions that follow.

Water use is measured in two ways: by withdrawal and by consumption. Withdrawal is the removal and transfer of water from its source to a point of use, such as a home, business, or industry. Most of the water that is withdrawn is eventually returned to its source. For example, much of the water used in industries and in homes is treated and returned to the river or lake it came from. When water is withdrawn and is not returned to its source, the water is consumed. For example, when a potted plant is watered, almost all of the water eventually enters the atmosphere by evapotranspiration through the leaves of the plant. The evaporated water was consumed because it was not directly returned to its source.

1. According to the passage, which of the following statements is true?
   a. Water that is consumed was never withdrawn.
   b. Water that is withdrawn cannot be consumed.
   c. A fraction of the water withdrawn is usually consumed.
   d. All of the water withdrawn is consumed.

2. Which phrase best describes the meaning of the term evapotranspiration?
   a. the absorption of water by plant leaves
   b. the process by which potted plants transpire their leaves by evaporation
   c. the process by which the atmosphere maintains water levels in plant leaves
   d. the process by which water evaporates from plant leaves

3. Which of the following statements is an example of consumption?
   a. A river is diverted to irrigate crops.
   b. A power plant takes in cool water from a lake and returns the water to the lake.
   c. A dam forms a reservoir on a river.
   d. An aquifer is recharged by surface water.

33. Analyzing Data How many gallons does Well B pump per day? What is the average pump rate for all of the wells? In one hour, how many more gallons of water will Well A pump than Well C?

34. Making Calculations If placing a container of water in your toilet tank reduces the amount of water per flush by 2 L, how much water would be saved each day if this were done in 80 million toilets? (Assume that each toilet is flushed five times per day.) Convert your answer into gallons (1 L = .26 Gal).

35. Communicating Main Ideas Why is water pollution a serious problem?

36. Writing Persuasively Write a letter to a senator in which you voice your support or criticism of a hypothetical water diversion project.

Now that you have read the chapter, take a moment to review your answers to the Reading Warm-Up questions in your EcoLog. If necessary, revise your answers.
Obiectives

- Construct a model of the Earth’s natural groundwater filtering system.
- **Using Scientific Methods** Test the ability of your groundwater filters to filter contaminants out of different solutions.

Materials

- beakers, 750 mL (5)
- glucose solution
- glucose test paper
- graduated cylinder
- gravel
- metric ruler
- soda bottles, 2 to 3L (4)
- red food coloring
- sand
- soil
- stirring rod
- wax pencil

Optional contaminants:
- cooking oil, detergent, fertilizer, vinegar, soda

Optional filter materials:
- alum, charcoal

Procedure

1. Label four beakers as follows: “Contaminant: glucose,” “Contaminant: soil,” “Contaminant: food coloring,” and “Water (control).”

2. Fill these beakers two-thirds full with clean tap water. Then add to each beaker the contaminant listed on its label. (The table on the next page shows how much of each contaminant you should use.) Stir each mixture thoroughly.

3. Copy the data table into your notebook. Carefully observe each beaker, and record your observations. Use some of the glucose test paper to test the glucose level in the glucose beaker.

4. Make four separate filtration systems similar to the one shown below. Your teacher will provide you with bottle caps that have holes poked through them. Fasten each cap to a bottle. Cut the bottom off of each soda bottle, and fill each bottle with layers of gravel, sand, and soil. Consider using the optional filter materials, such as alum or charcoal, but be sure to make each model identical to the next.

**Groundwater Filters**

As surface water travels downward through rock and soil, the water is filtered and purified. As a result, the water in aquifers is generally cleaner than surface water. In this investigation, you will work in small teams to explore how layers of the Earth act as a filter for groundwater. You will make models of the Earth’s natural filtration system and test them to see how well they filter various substances.
Observations of Substances in Surface Water

<table>
<thead>
<tr>
<th>Contaminant</th>
<th>Before filtration</th>
<th>After filtration</th>
</tr>
</thead>
<tbody>
<tr>
<td>Glucose (15 mL)</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Soil (15 mL)</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Food coloring (15 drops)</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Water (control)</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

5. You are now going to pour each mixture through a filtration system. But first predict how well the filters will clean each water sample. Write your predictions in your notebook.

6. Stir a contaminant mixture in its beaker, and immediately pour the mixture through a filtration system into a clean beaker. Observe the resulting “groundwater,” and record your observations in the table you created. CAUTION: Do not taste any of the substances you are testing.

7. Repeat this procedure for each mixture. Clean and relabel the contaminant beakers as you go along.

Analysis

1. **Analyzing Results** Test the glucose-water mixture for the presence of glucose. Can you see the glucose?

2. **Analyzing Results** Was the soil removed from the water by filtering? Was the food coloring removed? How do you know?

Conclusions

3. **Drawing Conclusions** How accurate were your initial predictions?

4. **Drawing Conclusions** What conclusions can you draw about the filtration model and the materials you used?

Extension

1. **Making Predictions** Choose a substance from the materials list that has not been tested. Predict what will happen if you mix this substance in the water supply.

2. **Evaluating Results** Now test your prediction. Use the filter that was the control in the earlier experiment. How did your results compare with your prediction?

3. **Analyzing Results** Compare your results with the results of other teams. What precautions do you recommend for keeping groundwater clean?
China’s Yangtze River is the third-longest river in the world after the Nile and the Amazon. The Yangtze River flows through the Three Gorges region of central China, which is famous for its natural beauty and historical sites. For thousands of years, the area’s sheer cliffs have inspired paintings and poems. This idyllic region seems like the sort of place that would be protected as a park or reserve. But in fact, it is the construction site for the Three Gorges Dam—the largest hydroelectric dam project in the world. When the Yangtze River is dammed, it will rise to form a reservoir that is 595 km (370 mi) long—as long as Lake Superior. In other words, the reservoir will be about as long as the distance between Los Angeles and San Francisco!

Benefits of the Dam

The dam has several purposes. It will control the water level of the Yangtze River to prevent flooding. About 1 million people died in the last century from flooding along the river. The damage caused by a severe flood in 1998 is estimated to cost as much as the entire dam project. The dam will also provide millions of people with hydroelectric power. China now burns air-polluting coal to meet 75 percent of the country’s energy needs. Engineers project that when the dam is completed, its turbines will provide enough electrical energy to power a city that is 10 times the size of Los Angeles, California. When the Yangtze’s flow is controlled, the river will be deep enough for large ships to navigate on it, so the dam will also increase trade in a relatively poor region of China.

Some Disadvantages

The project has many drawbacks, however. The reservoir behind the dam will flood an enormous area. Almost 2 million people living in the affected areas must be relocated—there are 13 cities and hundreds of villages in the area of the proposed reservoir. As the reservoir’s waters rise, they will also destroy fragile ecosystems and valuable archeological sites. Opponents of the project claim that the dam will increase pollution levels in the Yangtze River. Most of the cities and factories along the river dump untreated wastes directly into the water. Some people think the reservoir will become the world’s largest sewer when 1 billion tons of sewage flow into the reservoir every year.

Long-Term Concerns

People have also raised long-term concerns about the project. The dam is being built over a fault line. Scientists question whether the dam would be able to withstand earthquakes that may occur along the fault. If the dam burst, towns and
What Do You Think?

Hundreds of dams in the western United States provide electrical energy, drinking water, and food for crops, but the dams also flooded scenic canyons and destroyed ecosystems. Now that the environmental consequences of large dams are known, do you think that China should reconsider the Three Gorges Dam project?

Cities downstream would be flooded. Another concern is that the dam may quickly fill with sediment. The Yangtze picks up enormous amounts of yellowish soil and sediment as it flows across China. When the river is slowed by the dam, much of the silt will be deposited in the new reservoir. As sediment builds up behind the dam, the deposited sediment will reduce the size of the reservoir—limiting the flood-prevention capacity of the dam. In addition, productive farming regions below the dam will be deprived of the fertile sediment that is deposited every year when the river floods.

The enormous reservoir may also cause disease among the local population. The potential health risks include an increase in encephalitis and malaria. The most deadly disease spread by the Three Gorges Dam could be a parasitic disease called schistosomiasis.

Hidden Costs?

Supporters of the dam claim that the project will cost $25 billion, while opponents claim that the costs will be closer to $75 billion. The true cost of the dam may never be known because corruption and inefficiency have plagued the project from the start.

Controversy over the dam has prompted the U.S. government and the World Bank to withhold money for the project. Public opposition to the project has been silenced since the Tiananmen Square crackdown. But with help from private investment companies from the United States, the Chinese government is continuing with the project, and the dam is slowly being built. The world’s third-longest river will soon swell in the middle, and China will change along with it.